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SOLUTIONS MANUAL 3.
Apply How many atoms of hydrogen-1 are in a 1.007-g sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits. 1.007 g 1 H-1 atom $\underline{\quad}$ 1.672 10²⁴ g a. 6.023 10²³ H-1 atoms 4.

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Example 10.1 to

answer the question. •

The atomic number (Z)
= the number of

protons. • The mass

number (A) = the

number of protons +
the number of

neutrons. • Isotopes

are written with the

mass number to the

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upper left of the
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proceed by answering the Review Questions at the end of the chapter. This might also be a good time to read the Chapter Objectives, which precede the Review Questions. 1.09×10^4 kg P g P mol P mol P 40 10 ...

Chapter 10 Chemical Calculations and Equations

CHAPTER 10 REVIEW

States of Matter

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Assessment SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left.

b ionic crystal (a) has mobile electrons in the crystal
c covalent molecular crystal (b) is hard, brittle, and nonconducting
a metallic crystal (c) typically has the lowest melting point of the four

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with this answer sheet
to get ready for the
test on Chemistry of
Matter (Chapter 2 & 3)
Chemistry chapter 10
test review answers.

**Chemistry Chapter
10 Test Review
Answers**

Answer Is Not
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Answer Tolerances:

Yes: 09/29/19: 31

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accurate: Biology: Yes:

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